

SOLDERING OF ALUMINUM WITH ROSIN BASED AND INORGANIC FLUXES

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»Materials and Technologies for MEMS Packaging«



TAMURA ELSOLD: PRODUCT PORTFOLIO

TAMURA ELSOLD: The full range of soldering materials

- Solder
 - Bar, Ingot etc.
 - Wire (flux core, solid)
- Solder Pastes
- Fluxes

TAMURA Corp.

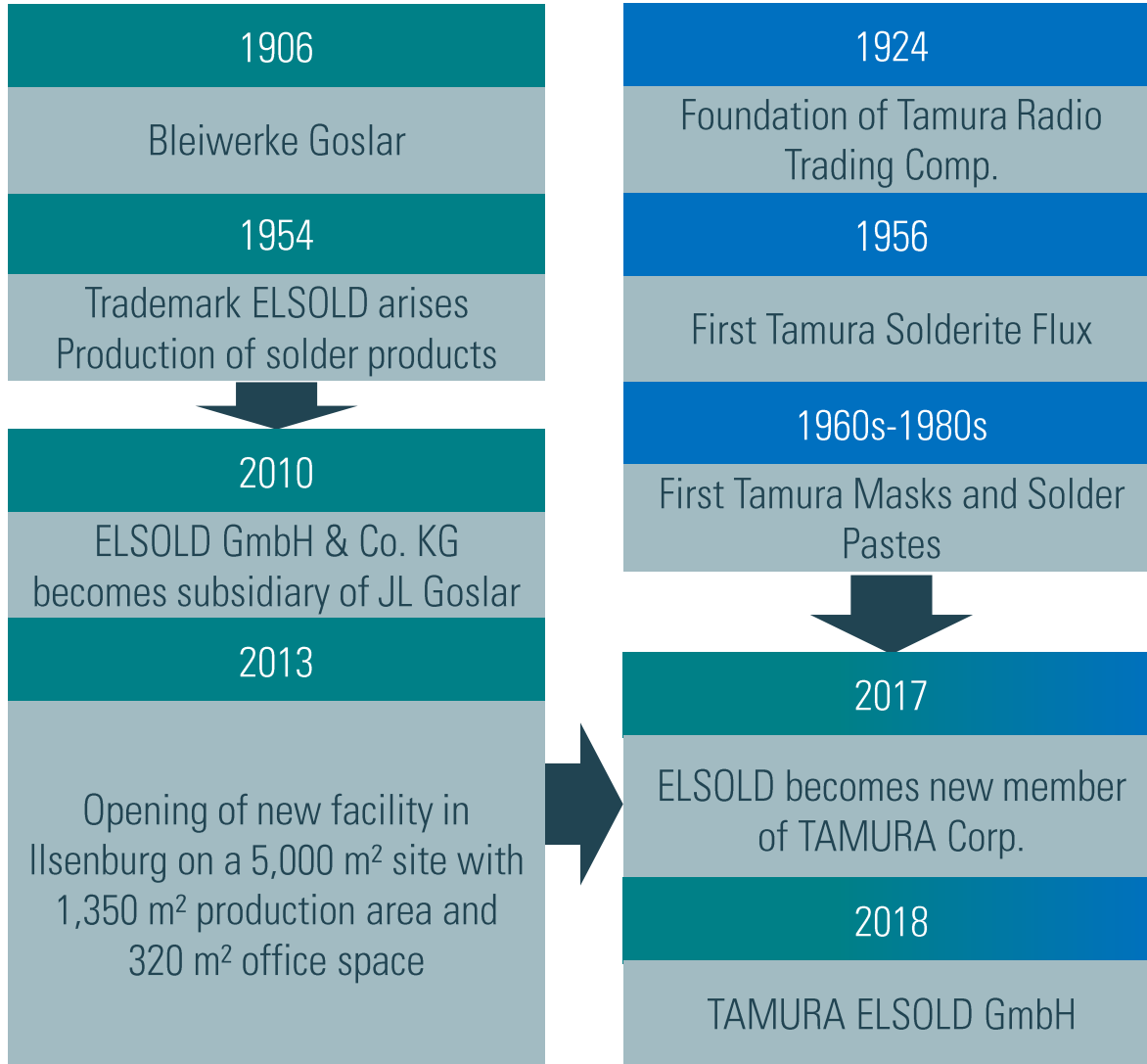
Electronic components

Electronic Chemicals & FA Systems

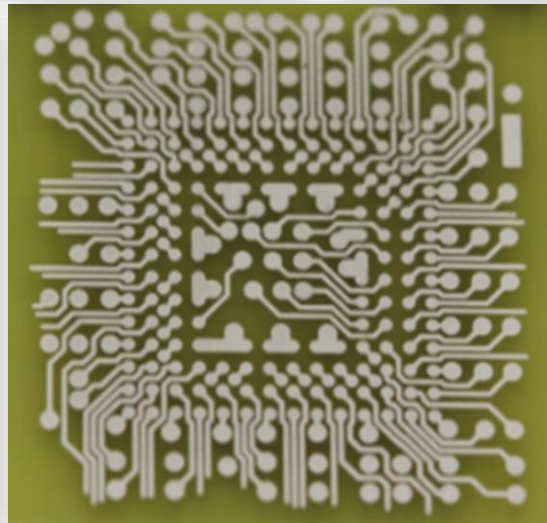
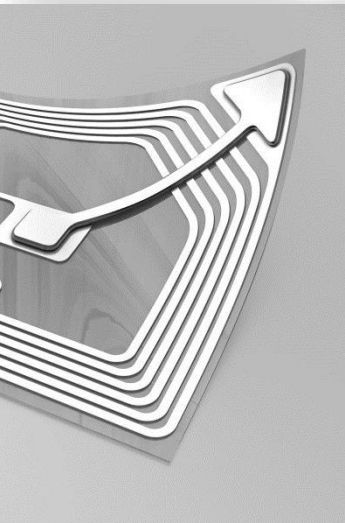
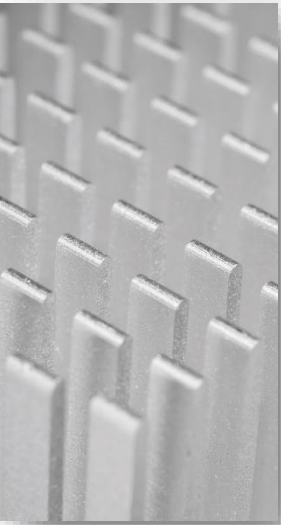
Information Equipment

- Solder Pastes
- Solder Fluxes
- Solder Resists

TAMURA ELSOLD: HISTORY



MOTIVATION: APPLICATIONS OF ALUMINUM FOR ELECTRONICS

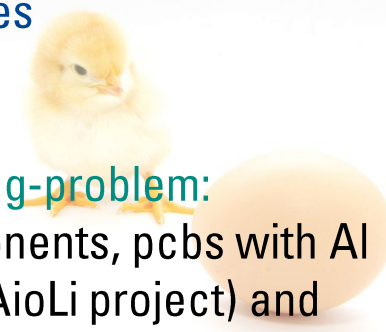


Current development: soldering of larger components and structures

Future: Macro → Micro

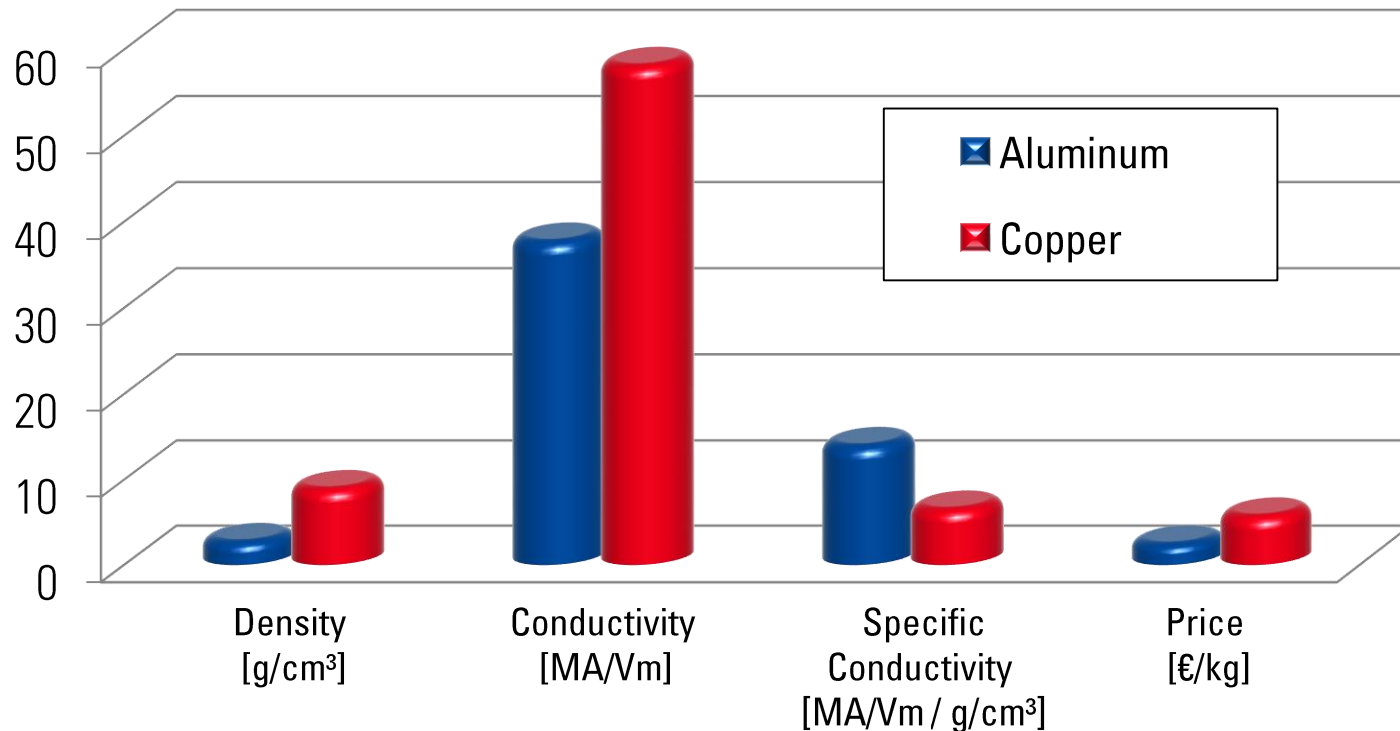
Closing gap & solving of chicken-and-egg-problem:

Development of Al-components, pcbs with Al conductive paths (ENAS AioLi project) and soldering technology /fluxes

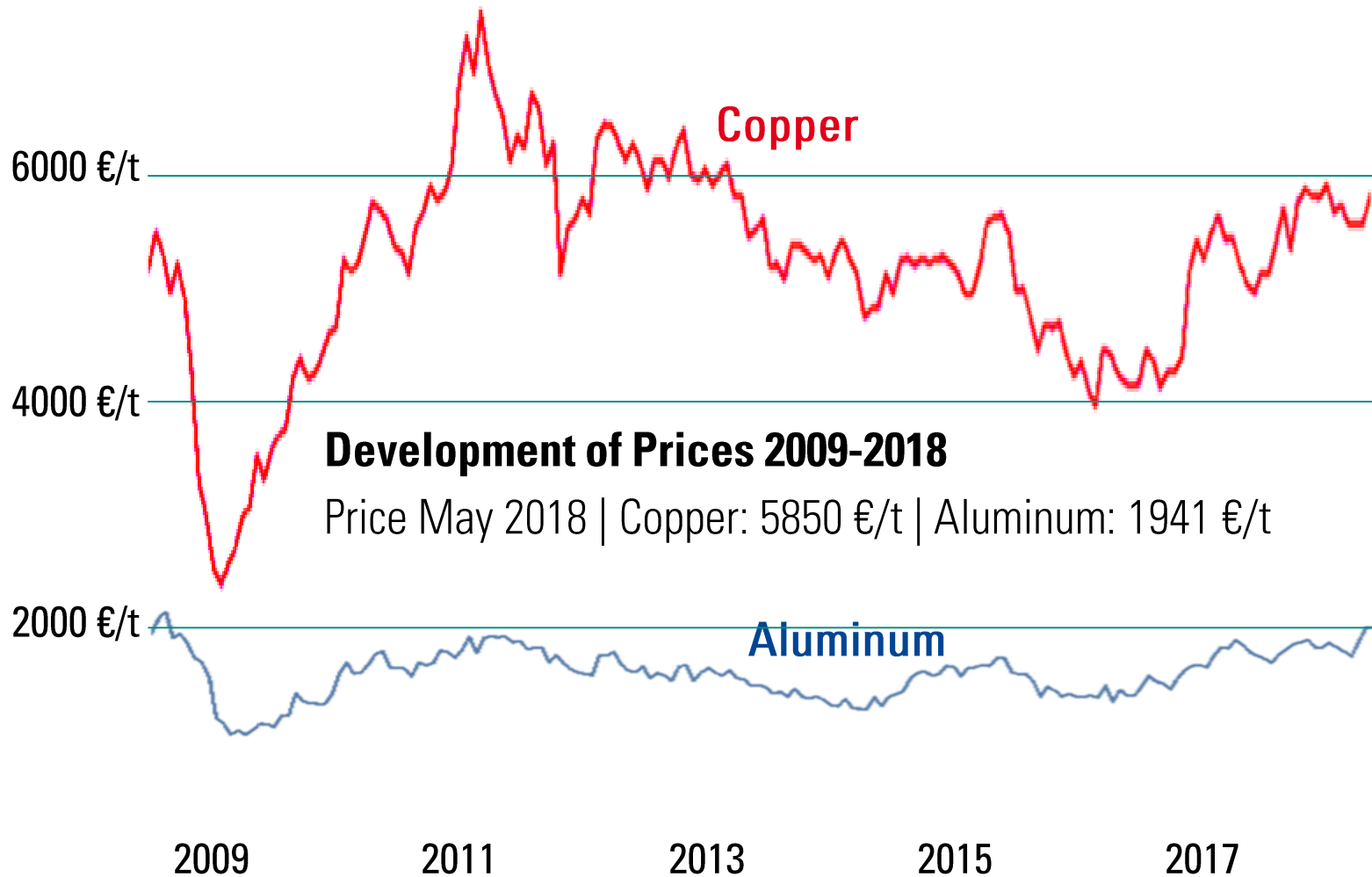


MOTIVATION: PROPERTIES OF ALUMINUM

- Up to 60% lighter than copper cables of comparable current carrying capacity
- Advantages at high-frequency technology due to skin effect:
 - Resistance factor 1.7 for DC
 - Resistance factor 1.7 for HF
- Advantages for transformers: Reduced springback at spool winding



MOTIVATION: PRICE OF ALUMINUM VS COPPER



MOTIVATION: REASONS FOR USE OF ALUMINUM CABLES



CHALLENGES OF ALUMINUM JOINING

Joining of Aluminum – Examples and Challenges

- Crimping – challenges: creep, only for specific geometries
- Brazing – challenges: Stable oxide layer, process temperatures too high for thermal sensitive components
- Soldering – challenge: stable oxide layer, non-removable with standard fluxes
 - Ultra Sonic Soldering/Bonding:
 - established for bond wires
 - special equipment and solder alloys necessary for soldering
 - **Soldering in standard processes, but with special fluxes**



Materials

- Pure Aluminum: good electrical properties, low strength
- Aluminum alloys: higher strength, more difficult to solder due to even more stable oxide layers (e.g. by MgO)

CHALLENGES OF ALUMINUM SOLDERING WITH FLUXES

Polylemma:

activity:

heavy reactive chemicals, e.g. fluorides, necessary to remove oxide layers

risk of corrosion:

residues of heavy reactive chemicals can cause strong corrosion

workability / storability:

heavy reactive chemicals may not react (inadequately) during production or storage. Limitation of usable chemicals, conventionally used for fluxes

applicability:

flux has to be applied in common processes, although typical substances cannot be used, e.g. to obtain good rheological properties

Flux types for soldering of Aluminum

- **Flux classification acc. to "old" DIN8511**
 - **F-LW-1: solder forming Tin- and Zinkchlorides (INH1):**
 - **Best activity [Down 1957]**
 - **High necessary soldering temperatures >400 °C**
 - **High corrosivity**
 - F-LW-2: organic compounds (ORH0)
 - F-LW-3: organic halogen compounds (ORH1)
- **Alternative: ROH1/ REH1 – improved protection against corrosion**

DEVELOPMENT OF FLUXES FOR SOLDERING ALUMINUM

Development of low melting INH1/F-SW-1-Flux

- Highest activity
- enlarged areas of application due to lower temperatures
- Flux paste (Al-P)

Development of rosin-based flux (ROH1)

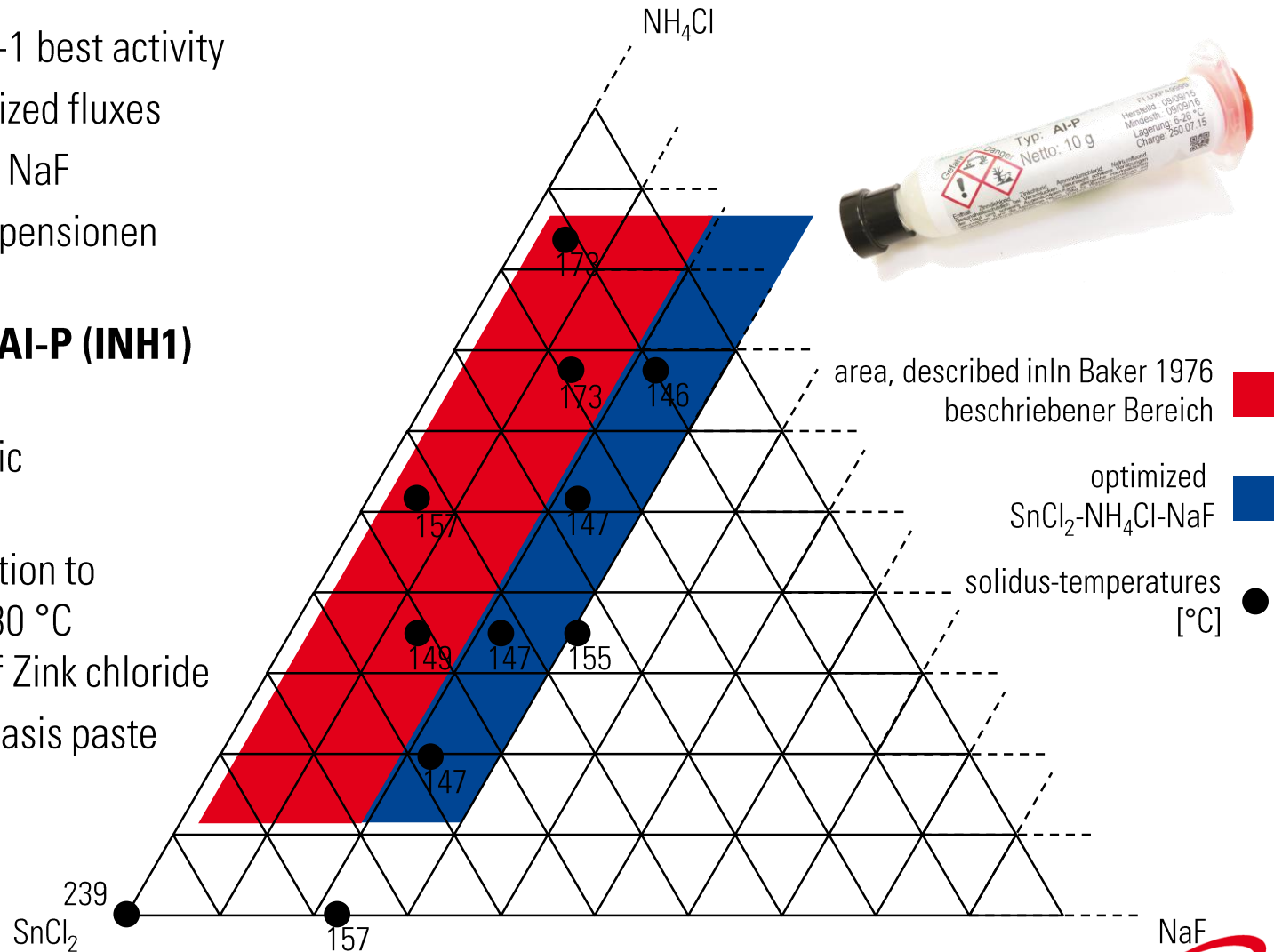
- Improved protection against corrosion
- Solid flux for flux cored solder wires (Al-S) and (based on same composition) as liquid flux (Al-L)

DEVELOPMENT OF AL-P (INH1)

- Down 1957: F-SW-1 best activity
- Baker 1976: optimized fluxes
 - SnCl_2 , NH_4Cl , NaF
 - Powder / Suspensionen

Development of Al-P (INH1)

- systematic thermoanalytic investigation
- Further reduction to lower than $130\text{ }^\circ\text{C}$ by addition of Zink chloride
- Use of inert basis paste



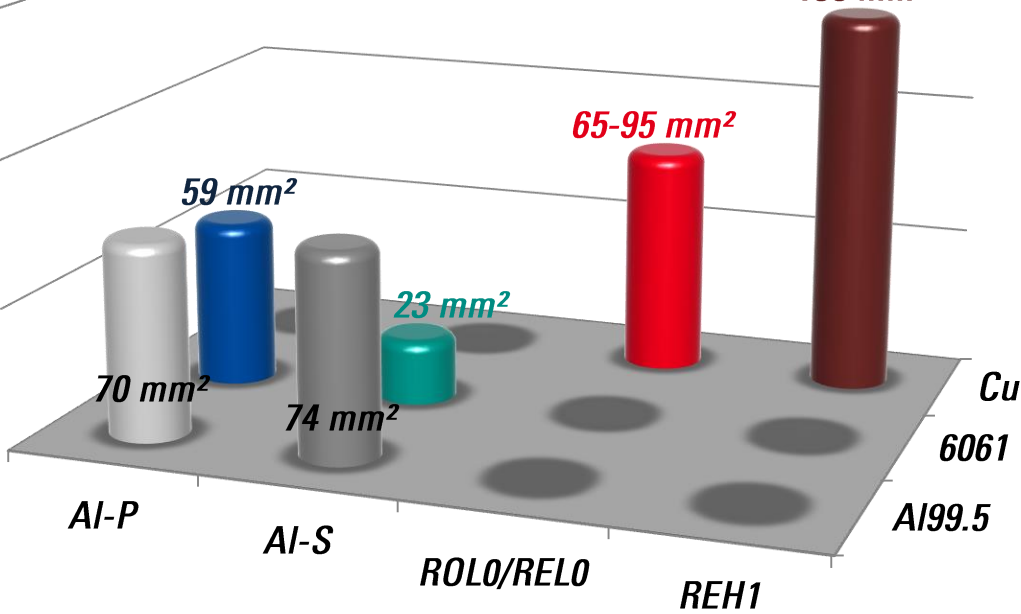
DEVELOPMENT OF AL-S/AL-L (ROH1)

- Well known encapsulating and protective effect of rosin on flux residues
- Miwa 1972: Aluminum flux with improved protection against corrosion
 - Curable epoxy resin & rosin
 - Heavy metal fluoroborates & triethanolamine
 - Usable as 2-component-system only, due to reactions of fluoroborates and rosin/resin
- **Development of Al-S/Al-L (ROH1)**
 - Basis: modified colophony
 - Activator: ammonium hydrogen difluoride
 - Special additives and production process to prevent inadequate reaction

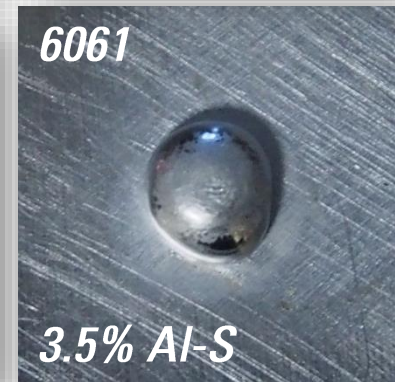
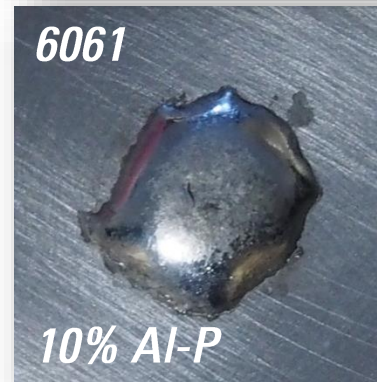
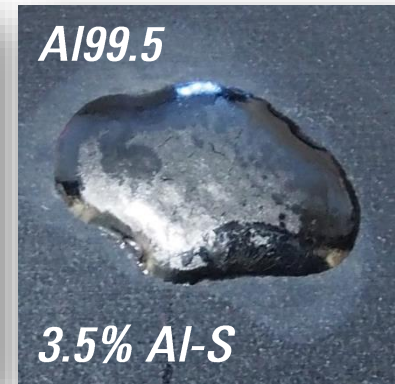
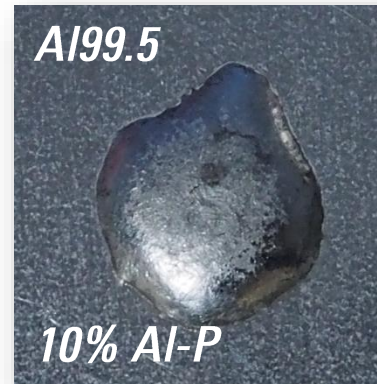
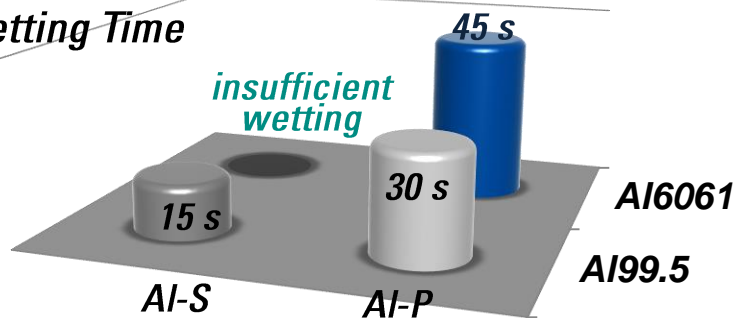


WETTING

Wetting Area



Wetting Time



- Solder: 0.4 g SAC305 combined with:
 - 10 % Al-P INH1 Flux Paste
 - 3.5% Al-S ROH1 Solid Flux
- Substrate: Al99.5 and 6061 (AlMg1SiCu)
- Soldering Temperature: 252 °C

WETTING

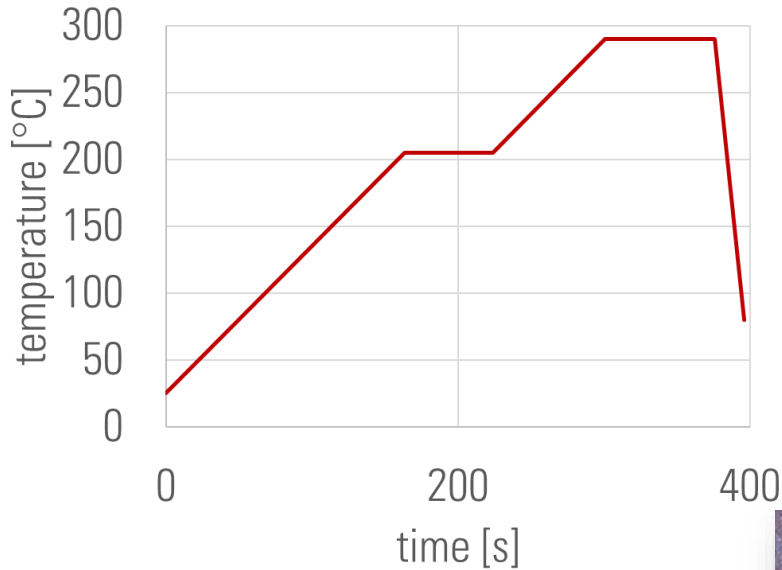
On pure Aluminum

- Good wetting with flux paste Al-P and flux cored wire Al-S
 - Wetting angle $< 30^\circ$
 - Similar to wetting with low activated flux on Copper → requirements of industrial joining processes are fulfilled generally
- Wetting slightly faster with Al-S than with Al-P
 - Solder forming reaction between flux Al-P and Aluminum needs more time than just oxide removing effect of fluoride activated rosin flux Al-S
 - But: application as paste → use in e.g. reflow processes
 - More time requirement is less critical than for iron soldering with cored wire

On alloy 6061

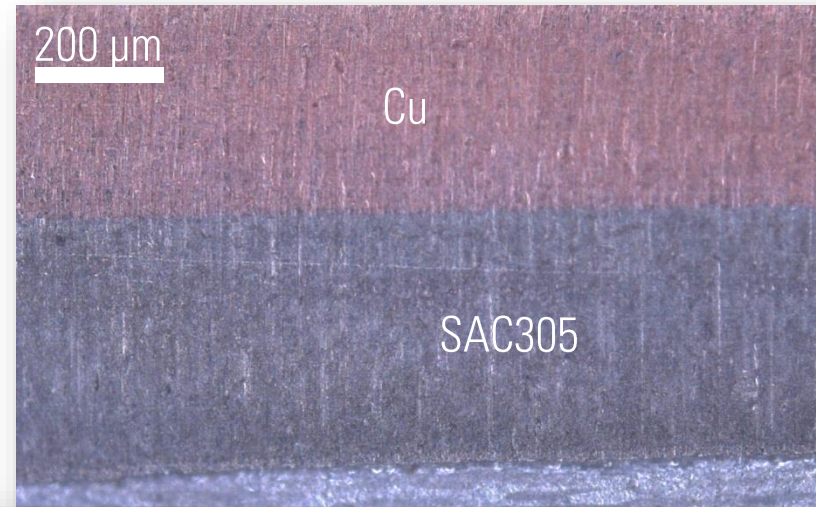
- Stable oxide layer due to Magnesium → High activity required
- Advantage of solder forming flux Al-P
- Still good wetting ($< 30^\circ$), but increased wetting time
- Al-S not suitable

REFLOW SOLDERING

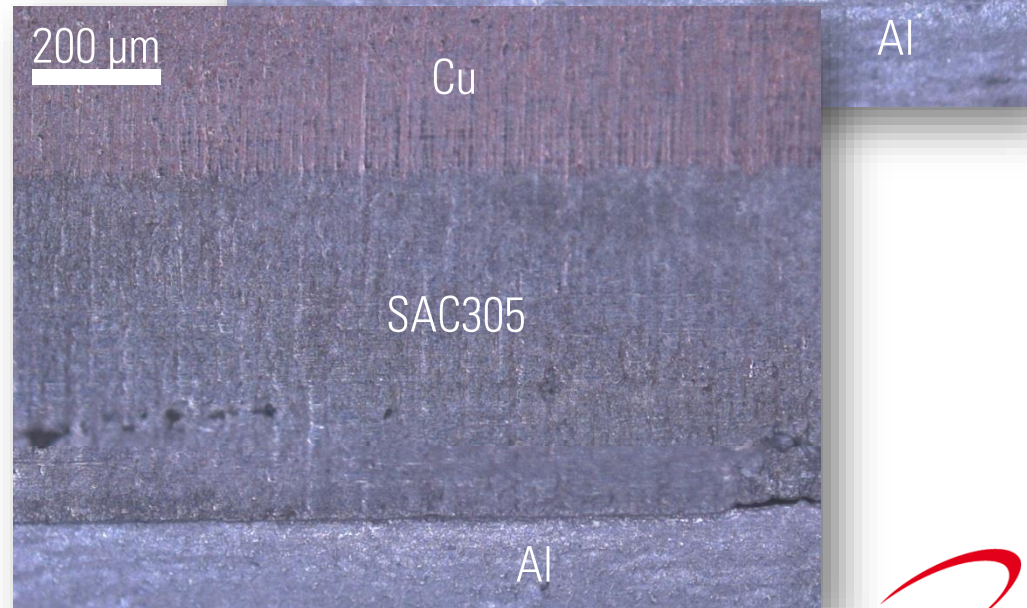


- Pure Copper to pure Aluminum
- SAC305
- Al-P (INH1, as flux paste + solid wire)
- Al-S (ROH1, as cored wire)
- Reflow

Al-P



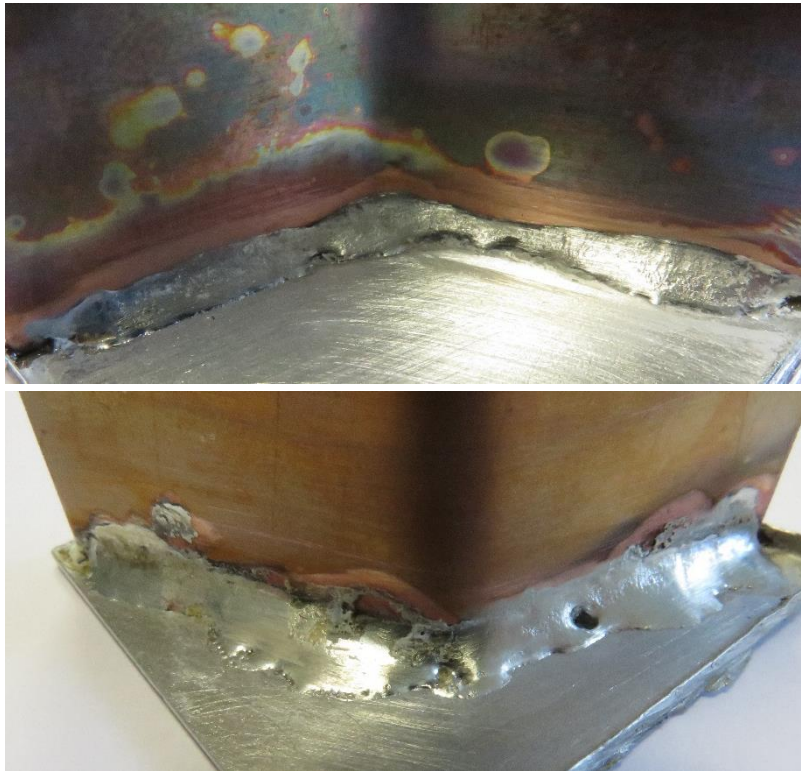
Al-S



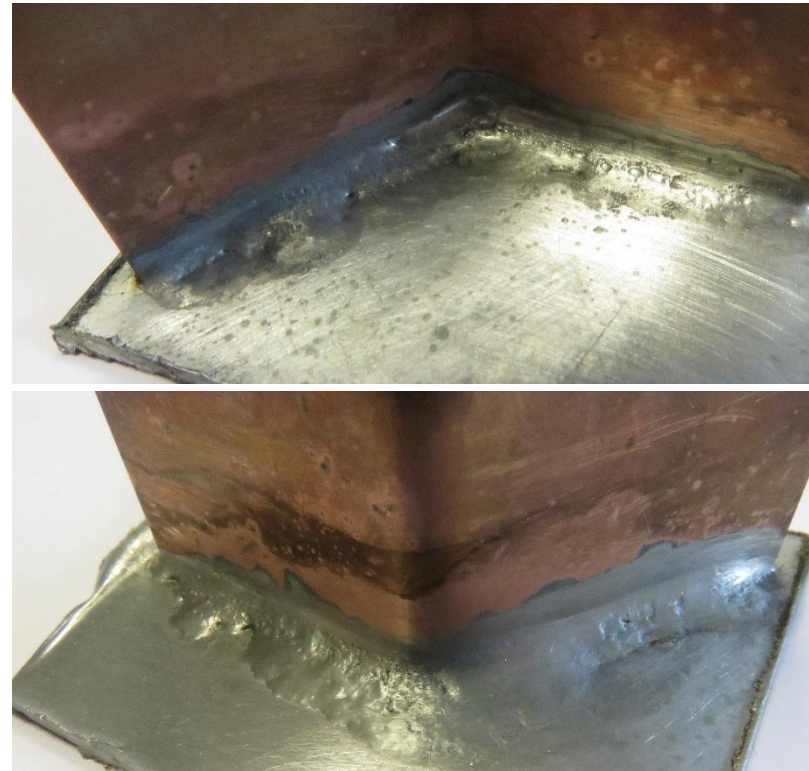
FLAME SOLDERING OF STRUCTURAL COMPONENTS

- Pure Copper to pure Aluminum // SAC305
- Al-S (ROH1, as cored wire): indirect flame heating
- Al-P (INH1, as flux paste + solid wire) direct flame heating

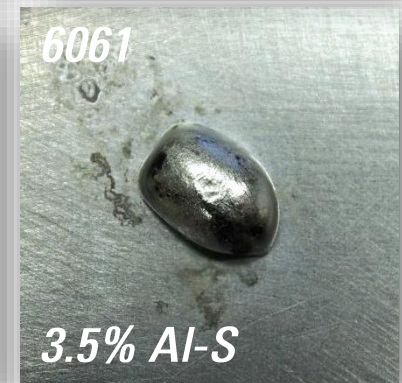
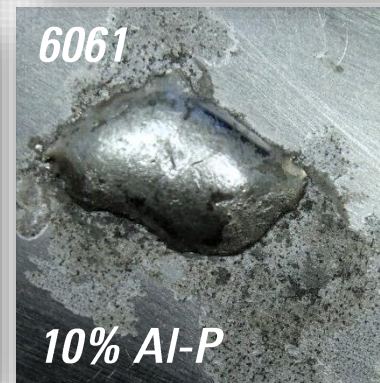
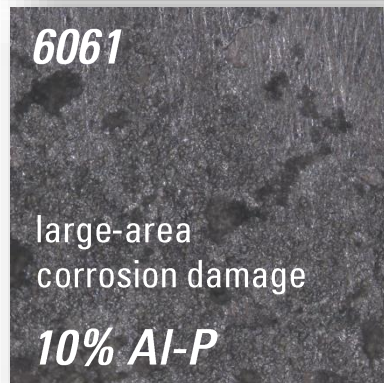
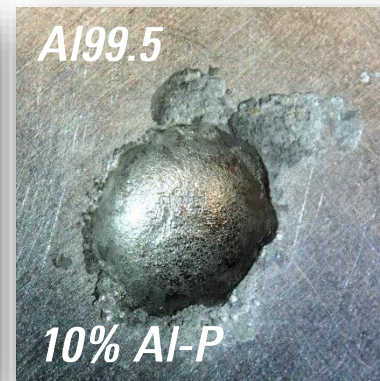
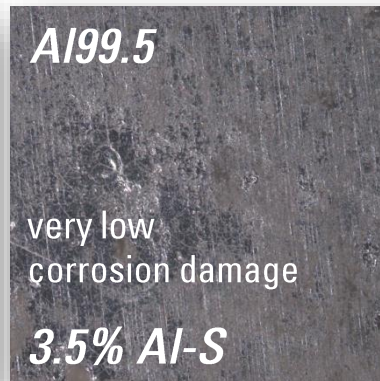
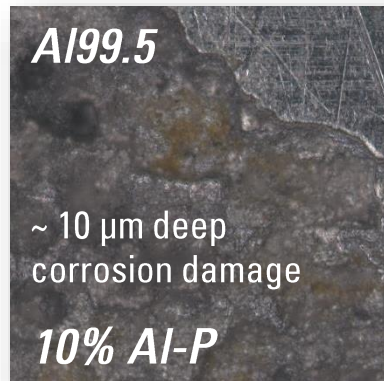
Al-S



Al-P



CORROSION BEHAVIOR



- Corrosion of wetting specimen. SAC305 combined with 10 % Al-P and 3.5% Al-S on Al99.5 and 6061; tested as soldered without cleaning, 7 days, 40 °C and 93% r. H.; cleaned after corrosion test
- **Cleaning necessary for INH1/F-SW-1** flux paste Al-P
- **Cleaning suggested for ROH1** flux Al-S, but significantly **less critical**.

CONCLUSION & OUTLOOK

- Aluminum can be soldered by use of ROH1 and INH1 fluxes
- Both Al-Al and Al-Cu joints are possible
- INH1 fluxes offer highest activity, also for soldering of Al alloys, available as flux paste AI-P
- ROH1 fluxes offer lower risk of corrosion by protection layer of rosin, available as solid flux AI-S for flux cored solder wires and liquid flux AI-L
- Different soldering processes for bigger structures were done successful
- Outlook: Adaption to micro level. E.g. soldering of pcb with Al conductive paths. Parallel development of optimized fluxes and Al components, supporting each other?!