

Materials for LED packaging

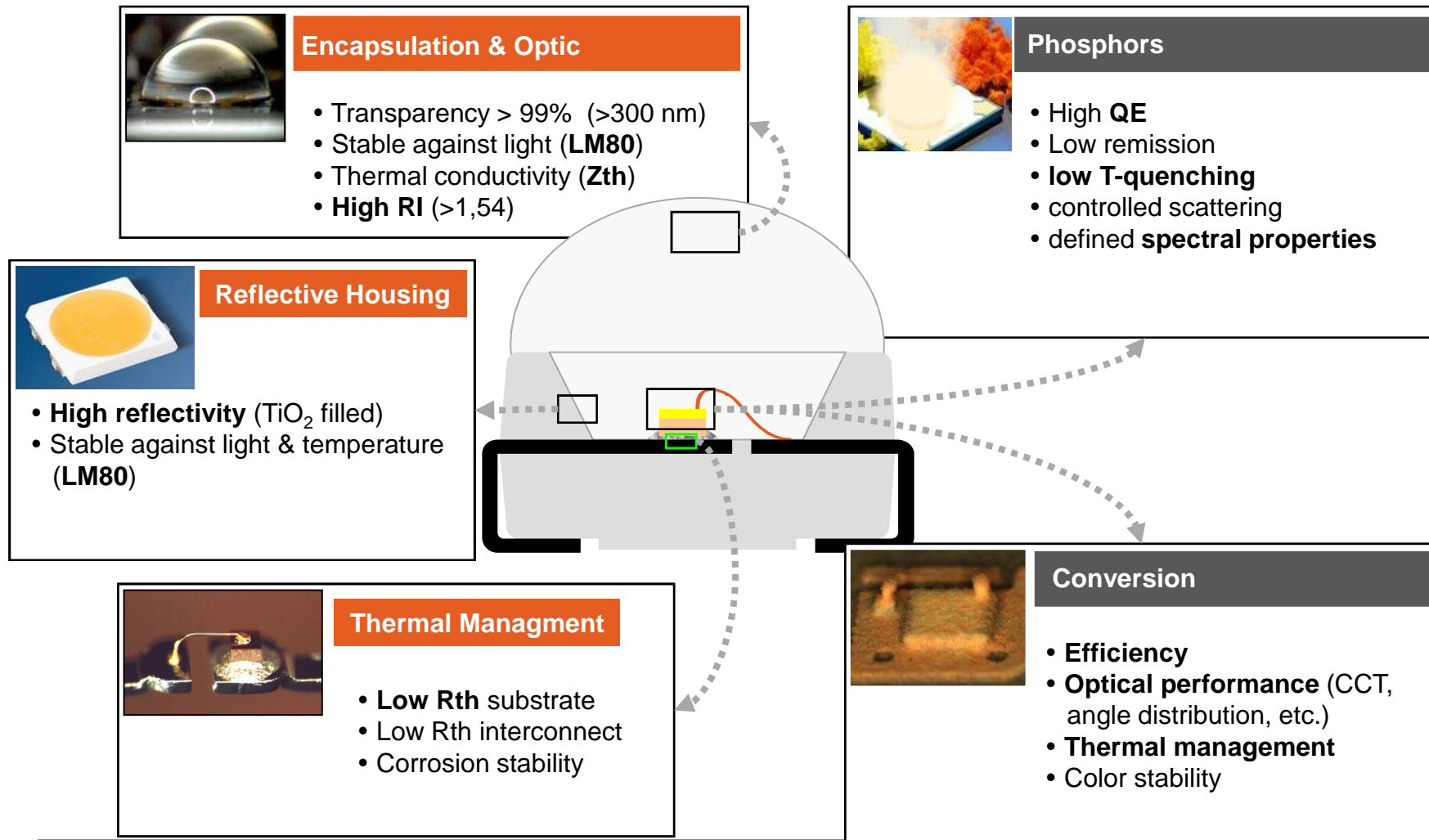
Dr. N. von Malm | June 2018 | Chemnitz

Agenda

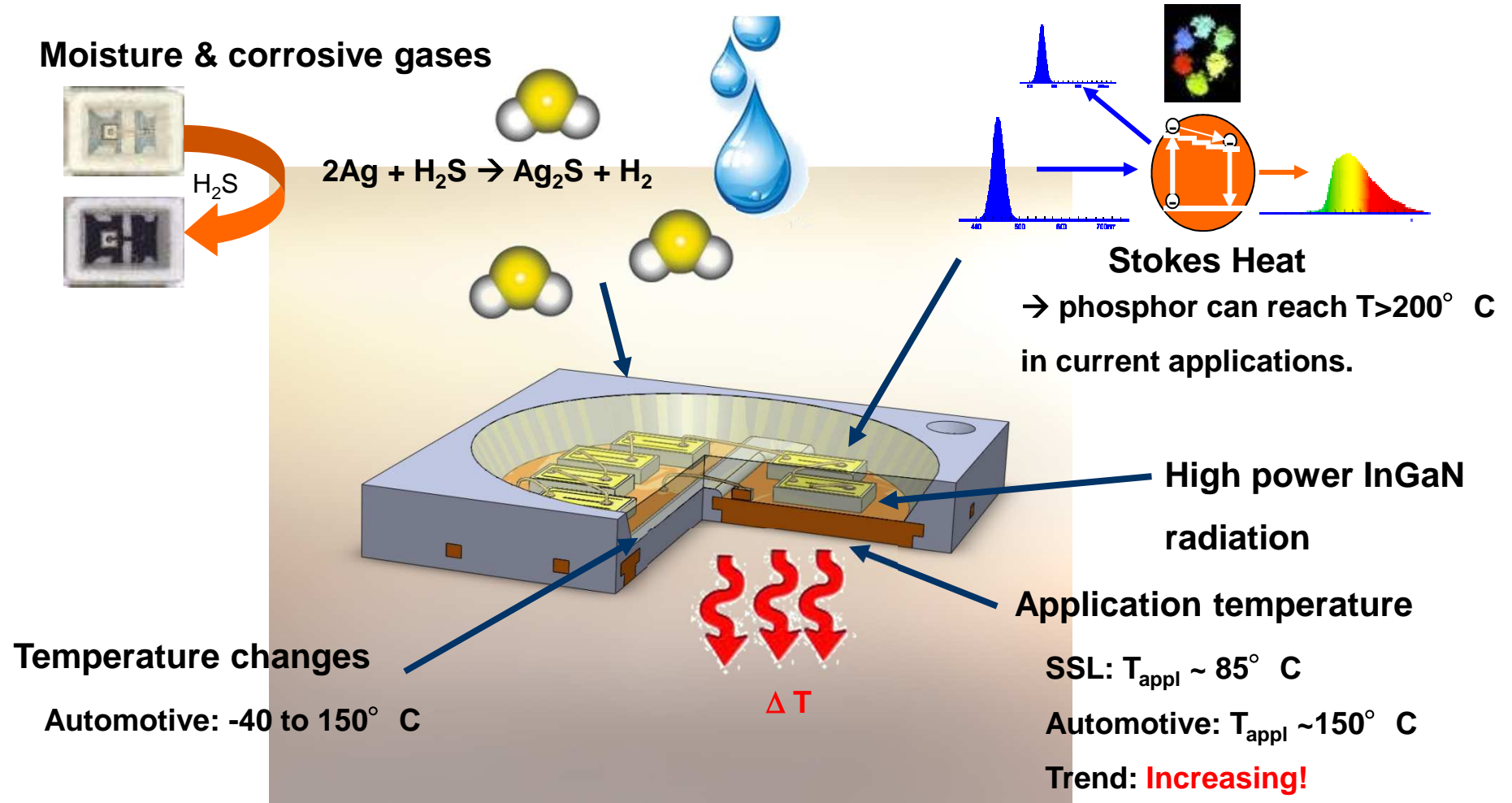
1. Introduction: Materials for LEDs
 2. Packaging Materials
 3. Materials for color conversion (Phosphors)
-

Advanced materials for pcLEDs

What makes the difference?



Today's challenges for a LED in modern environment...

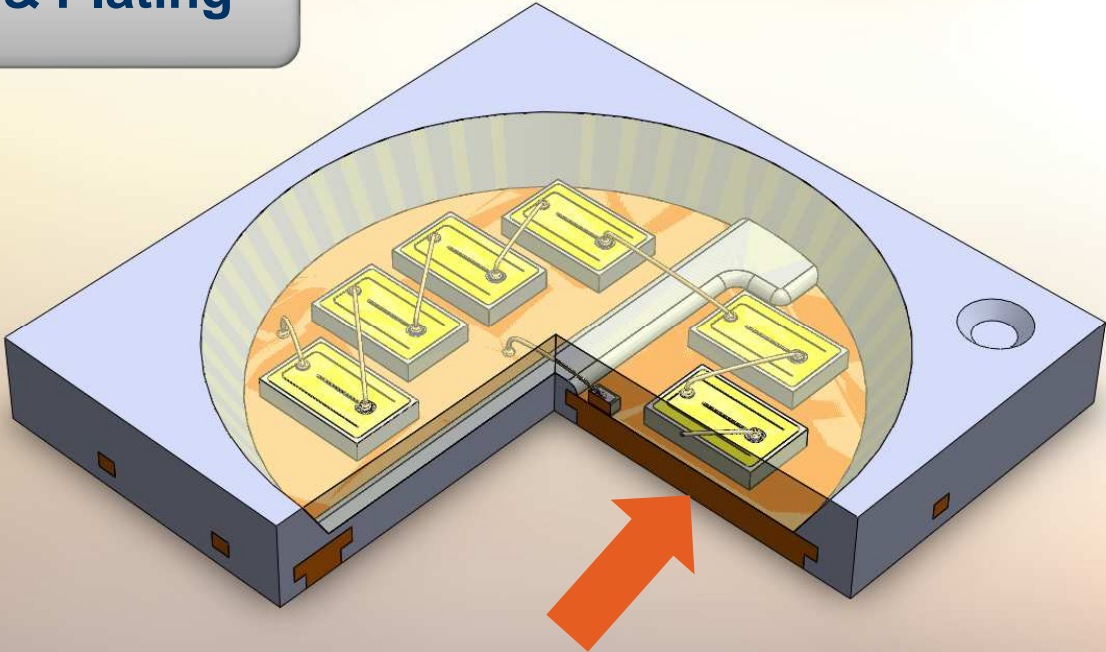


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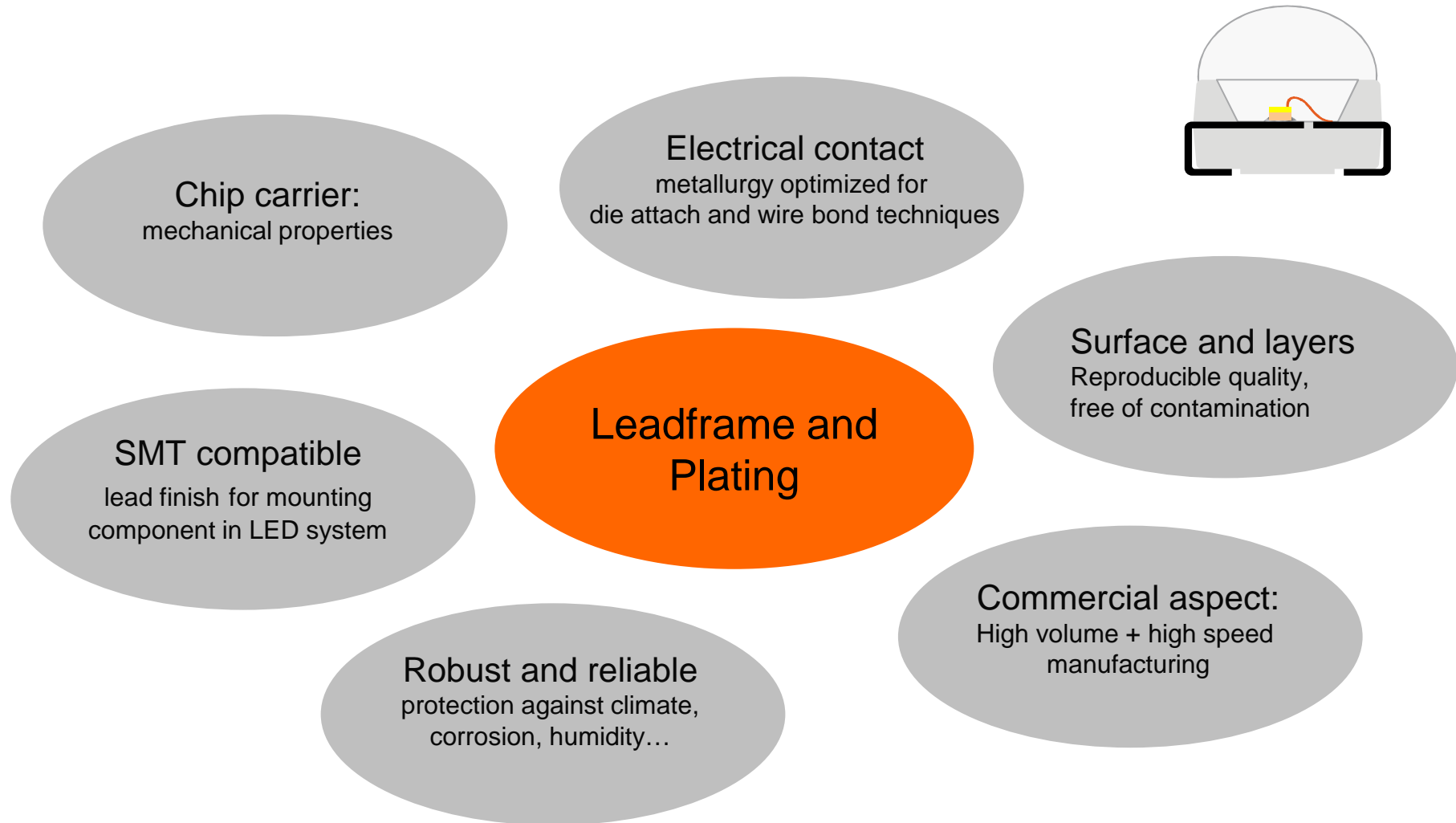
Leadframe and Plating

Leadframe & Plating



Leadframe and Plating

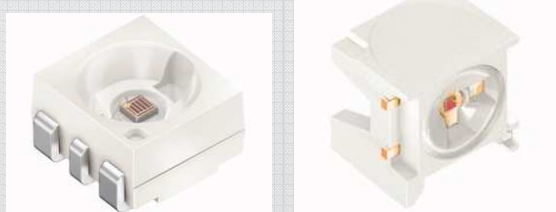
Requirements for Platings



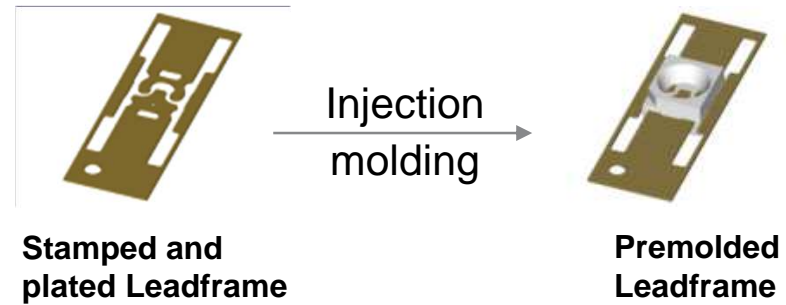
Leadframe and Plating

Type of Package Substrates


Premolded leadframe



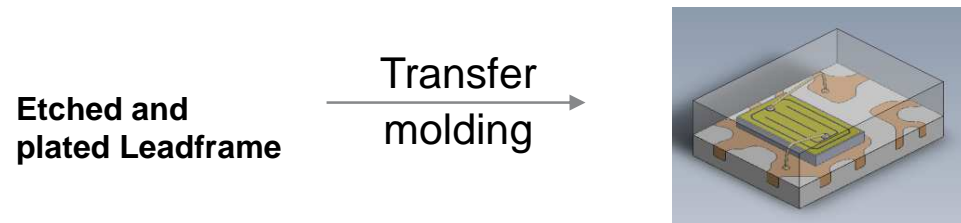
Advanced Power TOPLED Power SIDELED



Quad flat no lead (QFN)

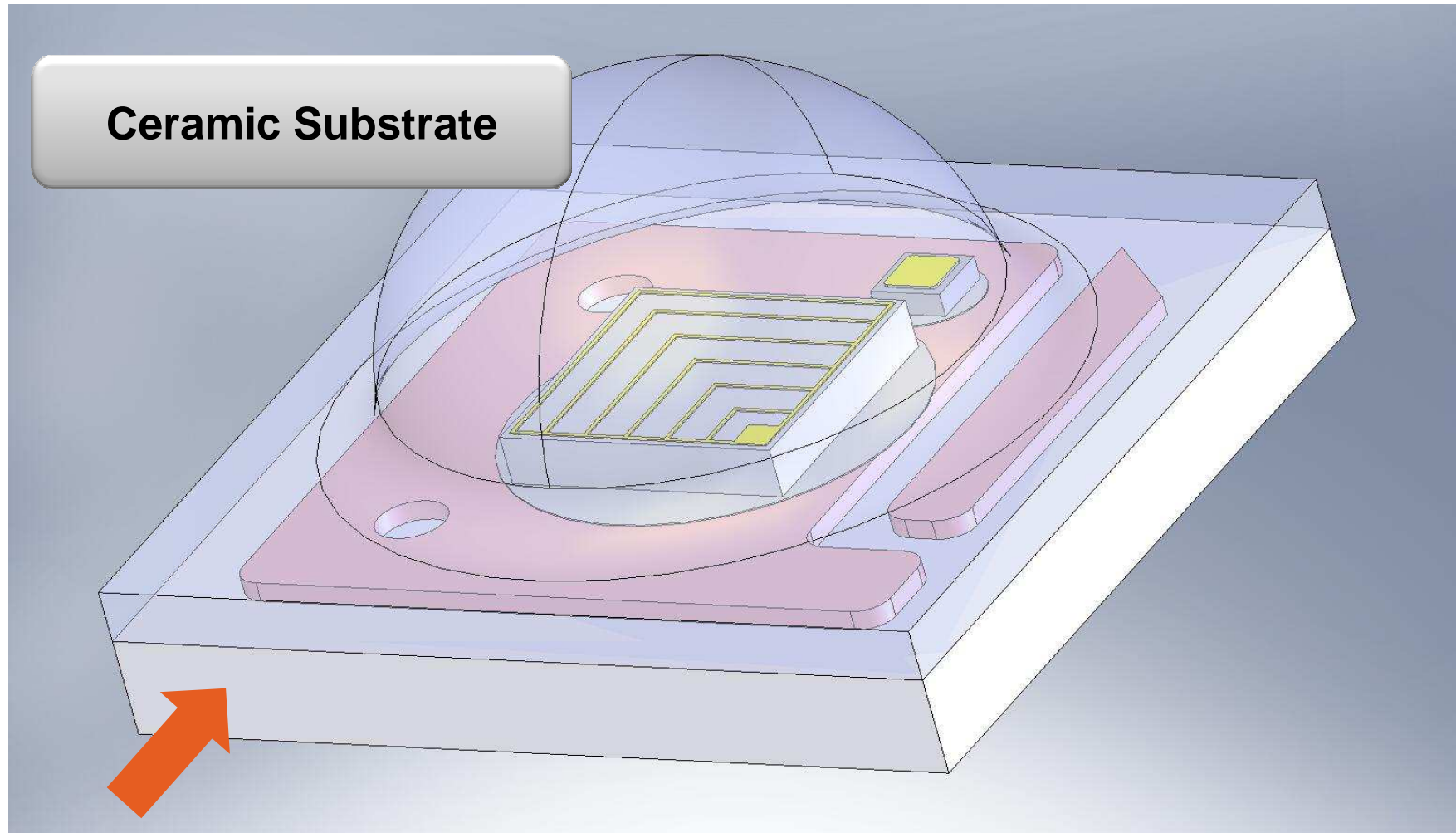


OSRON Black Flat Duris S5 (3030)



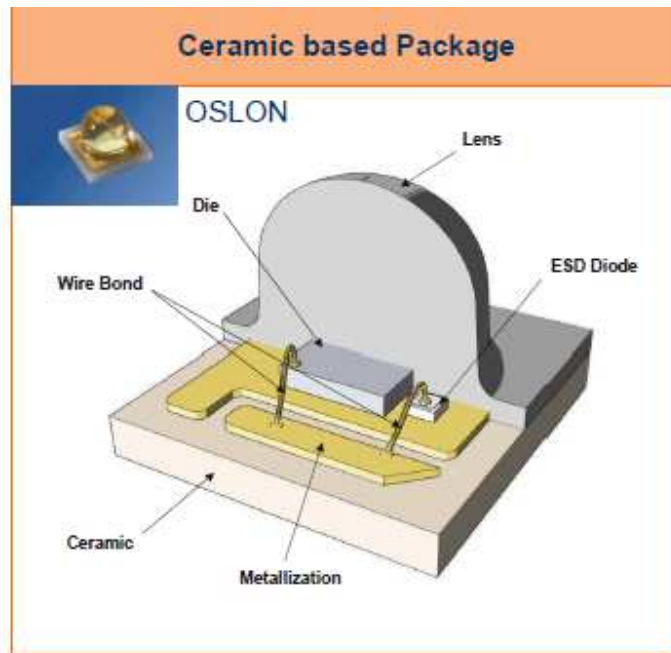
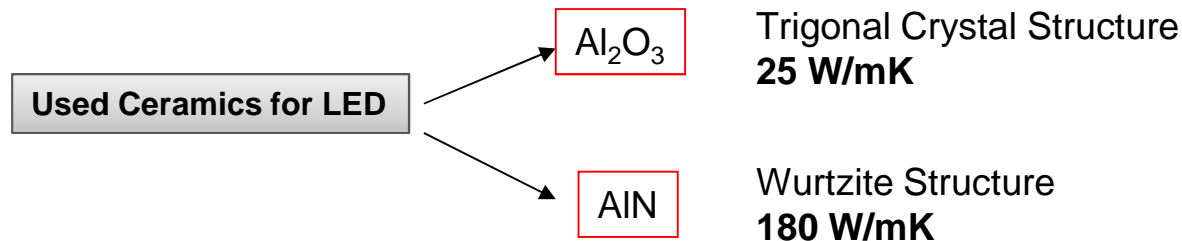
Electro-plating: Cu/Ni/Pd/Ag or Cu/Ni/Pd/Au

Ceramic Substrates



Ceramic Substrates

Material Basic Functions

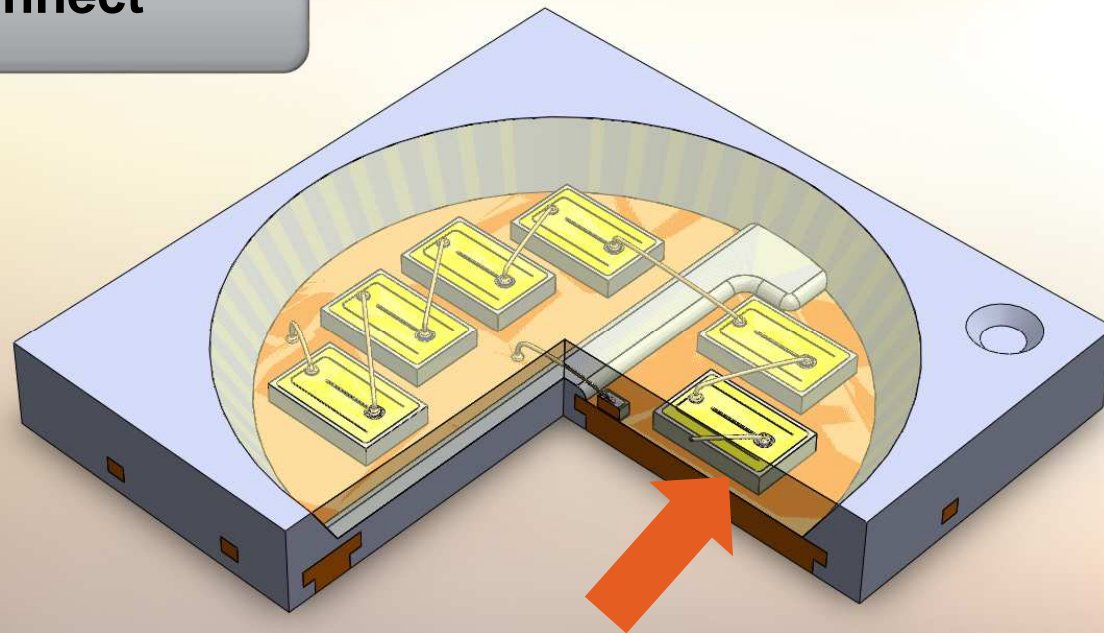


Ceramic properties :

- **Base substrate** (Metallization area Ag or Au).
- Electrical **insulation**: insulated backside
- High **thermal conductivity**: dissipating heat in high Power LEDs.
- **Thermal stability**: Long term reliability at high temperature.
- Similar **expansion coefficient**: ceramic substrate 5-8ppm/K, LED chip 5-6ppm/K
- **Reflective** surface for LED chip

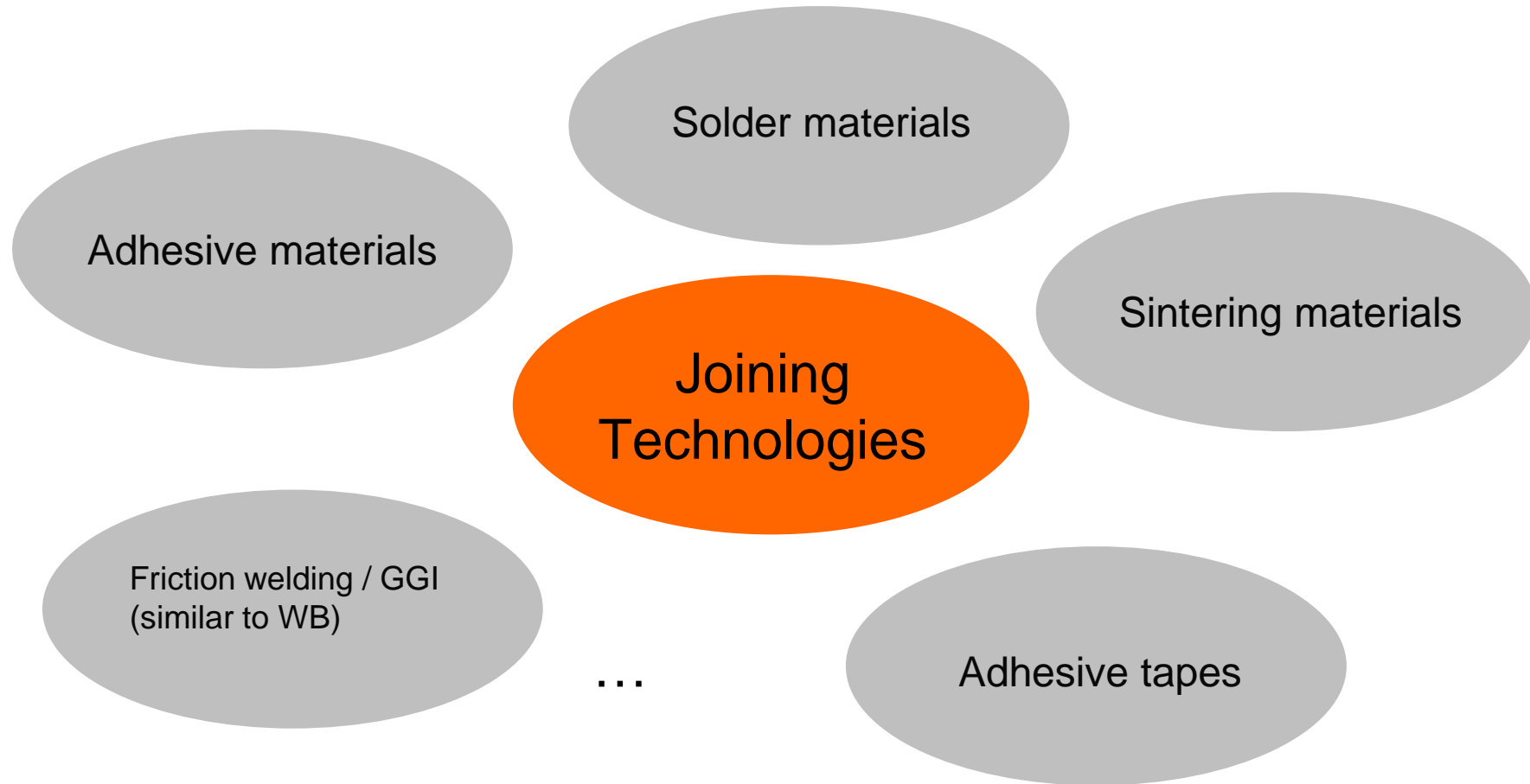
Interconnect Materials

Interconnect



Interconnect Materials

Joining Technologies & Material



Interconnect Materials

Materials for die attach

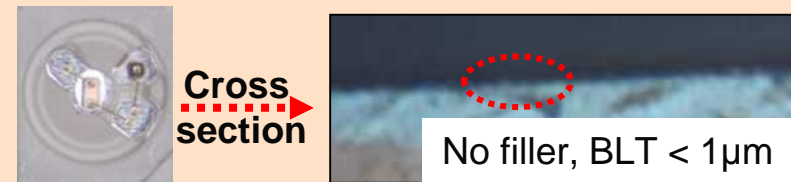
Electrically conductive glues λ 1-7 W/mK

- Adhesives for die attach: LED, ESD
- Adhesives for submount & heatsink attach



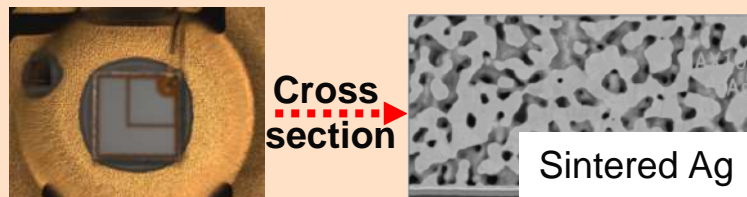
Electrically non-conductive glues \sim 0.1 W/mK

- Adhesives for sapphire die attach (transparent)
- Adhesives for optics attach: Lens, reflector



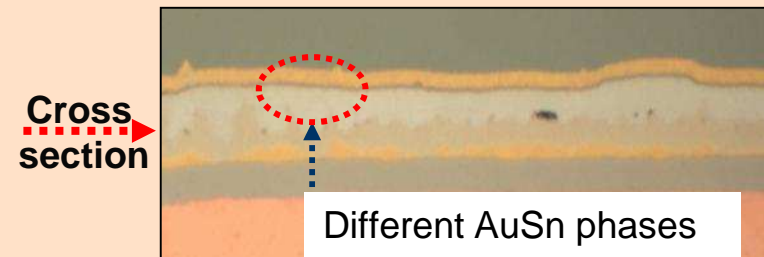
Sintering material λ 50-100 W/mK

- Sintering material for die attach
- Sintering for submount & heatsink attach



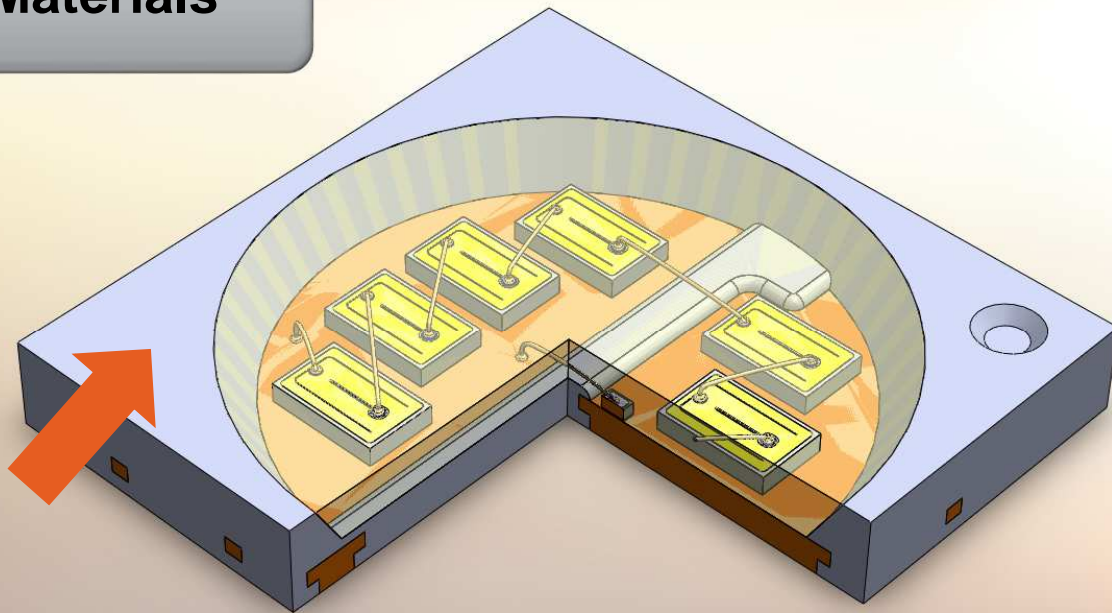
Solder material λ 40-60 W/mK

- Solder materials for die attach e.g. AuSn



Housing Materials

Housing Materials

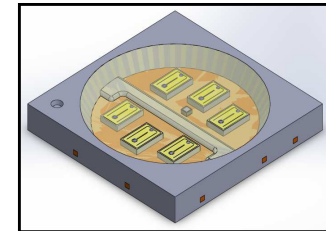
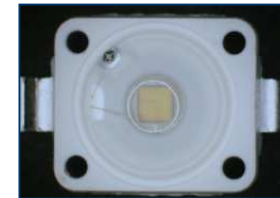


Housing Materials

Basics

Requirements for housing materials:

- high reflectivity > 90 % at 440 nm
- superior heat and light stability
- crack resistance and good adhesion (LFr, encapsulant)
- resistance to solder heat (>260°C)



Used material classes:

Semi-crystalline thermoplastics:

- PPA (PA 6T/6I, PA9T PA 10T), PCT, LCP, PEEK, PEI

Thermosettings:

- white or black Epoxy (EMC, 1-component frozen product)
- epoxy resin with anhydride hardener
- high dimensional, mechanical and environmental strength

Silicones:

- high mechanical, heat and light durability
- high reflectivity

Hybrids

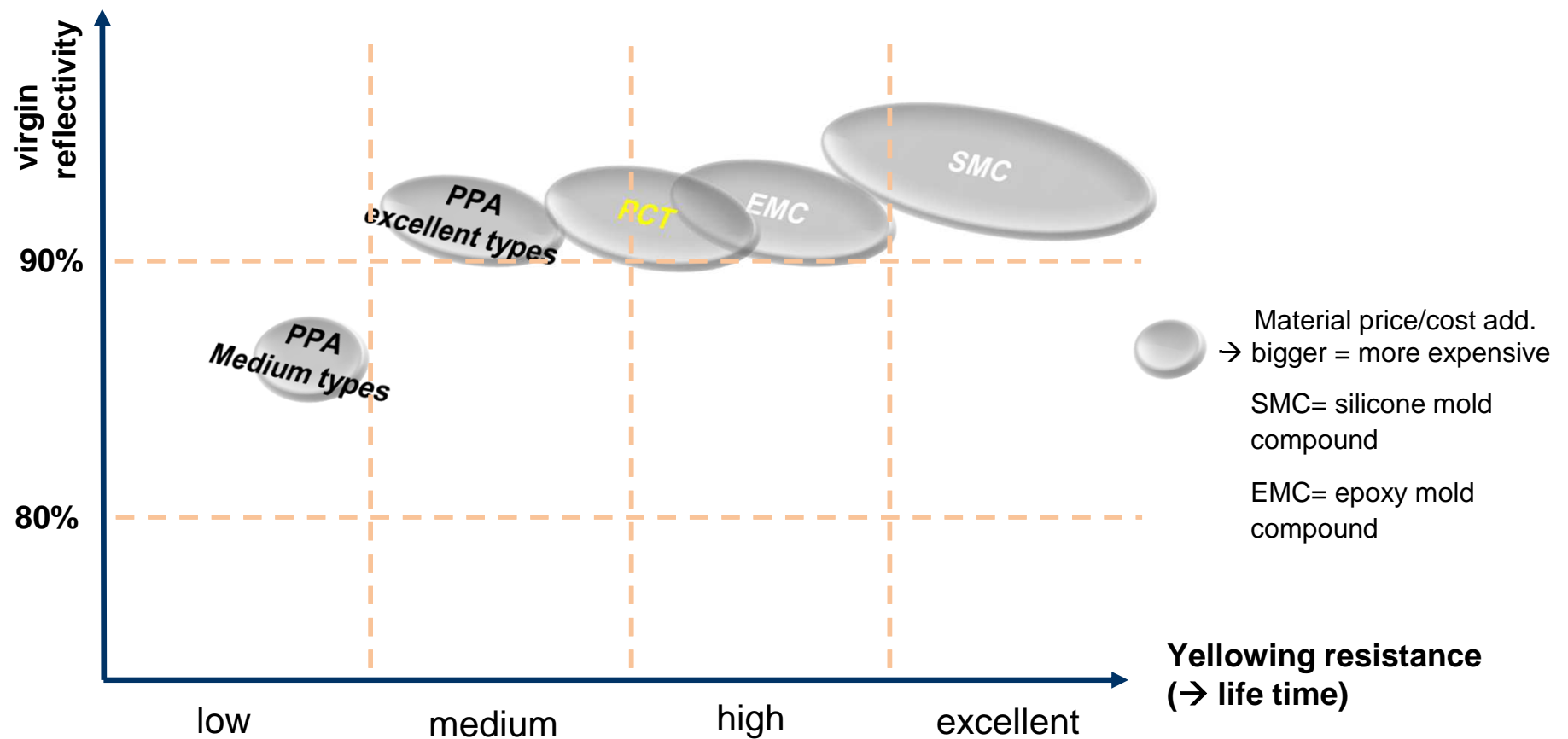
- epoxy network with silicone structures
- silicone network with epoxy structures
- high heat and light resistance

Fillers

- glass fibers
- pigments
- SiO₂
- TiO₂
- other mineral fillers

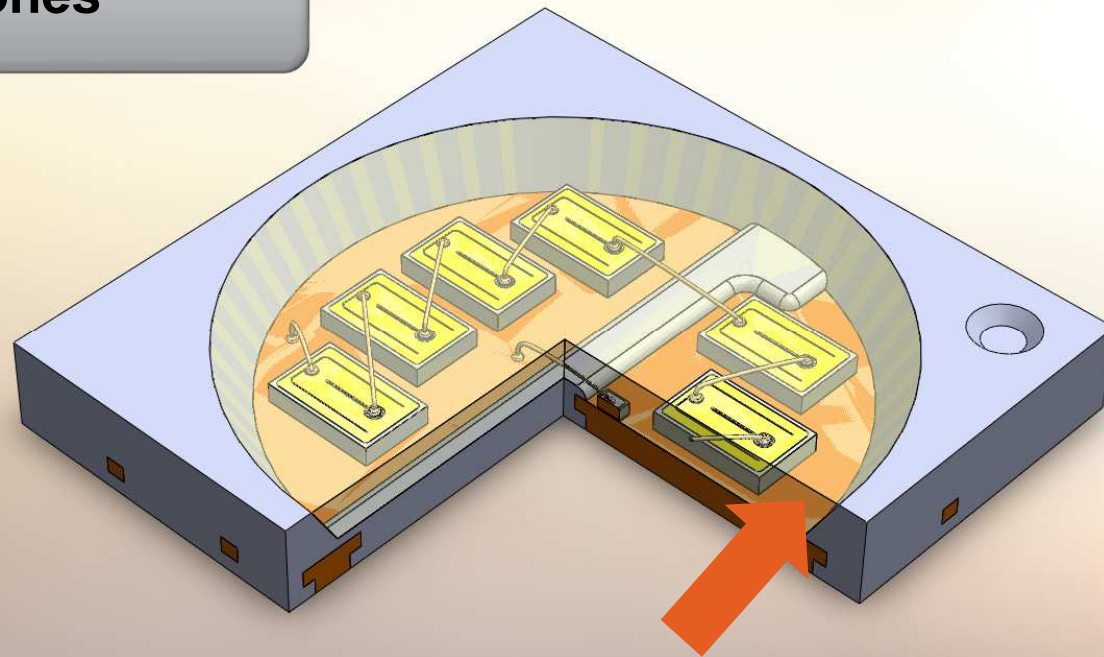
Housing Materials Comparison

Proper selection of adequate material stronger demanded since more and better materials are available!



Encapsulation Materials

Silicones



Advantages of using silicone as an encapsulant

Elastomer

Radiation Resistance

Chemical Resistance

Water Proof

Thermal Stability

Transparency

chip

Thermo Oxidative Resistance

chemical structure:

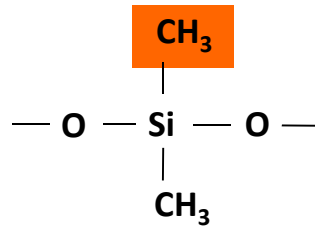
$$\begin{array}{ccccccc}
 & R1 & & R1 & & R1 & \\
 & | & & | & & | & \\
 - & O - Si - O - & Si - O - & Si - O - & & & \\
 & | & & | & & | & \\
 & R2 & & R2 & & R2 & \\
 & & & \left. \vphantom{\begin{array}{c} R1 \\ | \\ -O-Si-O- \\ | \\ R2 \end{array}} \right\} n & & &
 \end{array}$$

bond energy:

C - C	345 kJ/mol
Si - O	452 kJ/mol

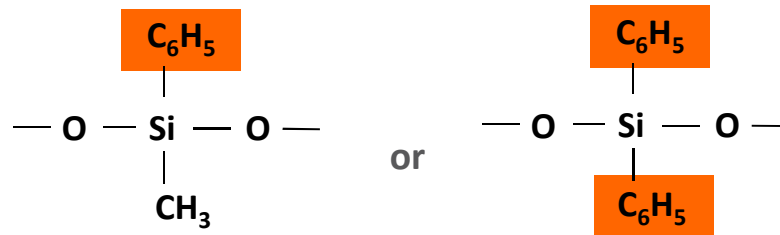
Types of silicones

Methyl-Silicones:



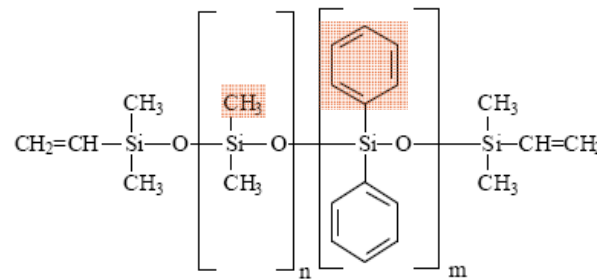
Methyl side group
low refractive index (LRI)

Phenyl-Silicones:



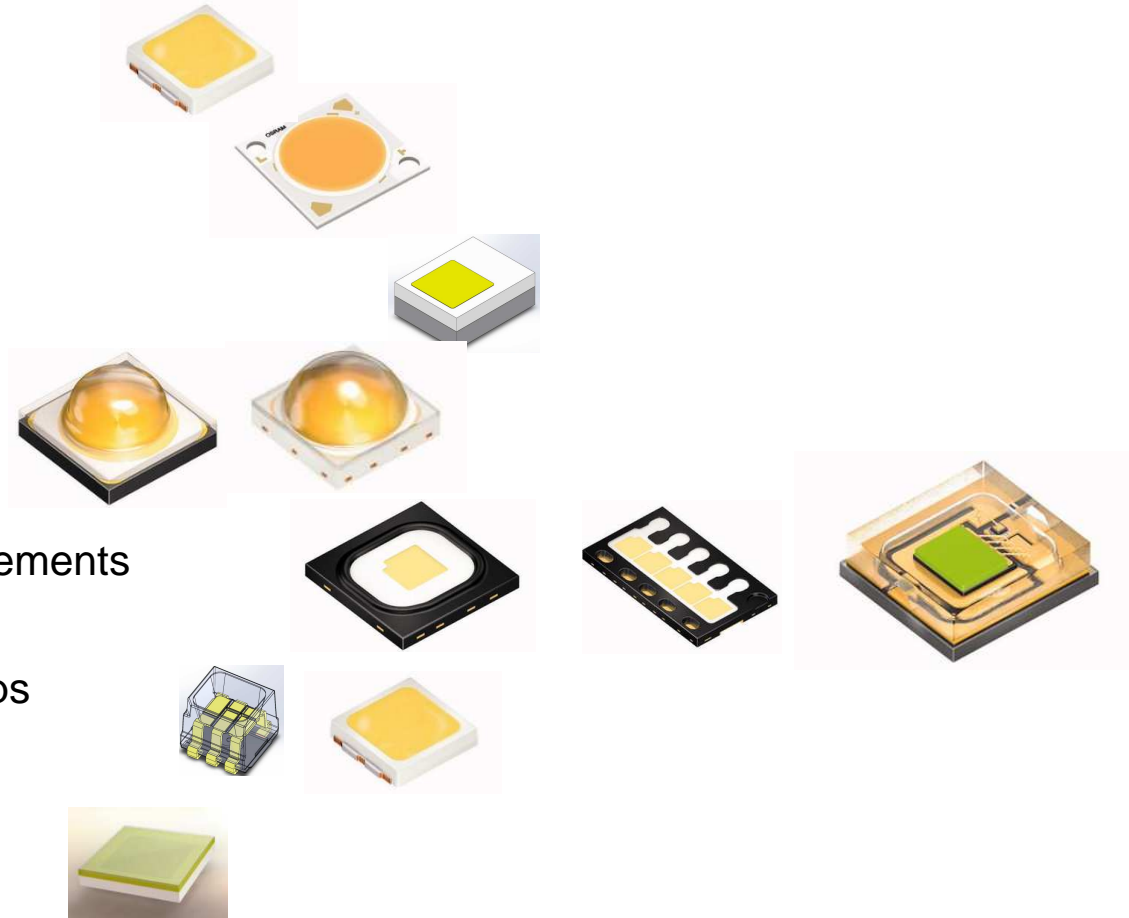
Phenyl side group
high refractive index (HRI)

actual polymer composition:



Applications for silicones

- Volume conversion
- Sedimentation
- TiO_2 matrix
- Lens material
- Glue for conversion elements
- Glue for Sapphire Chips
- Phosphor sheets



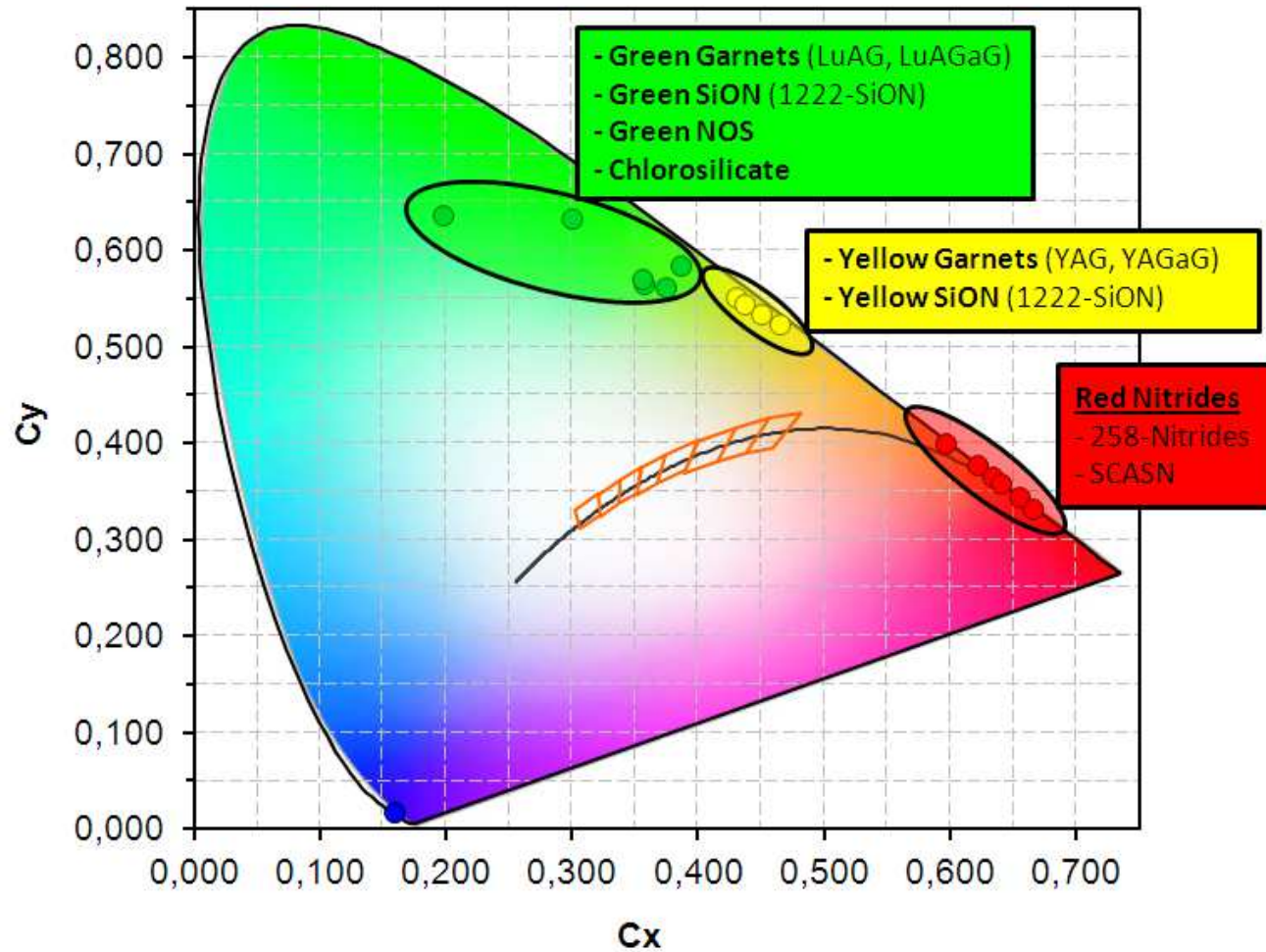
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Phosphor Materials

Chemistry – Material Classes



Phosphor Materials

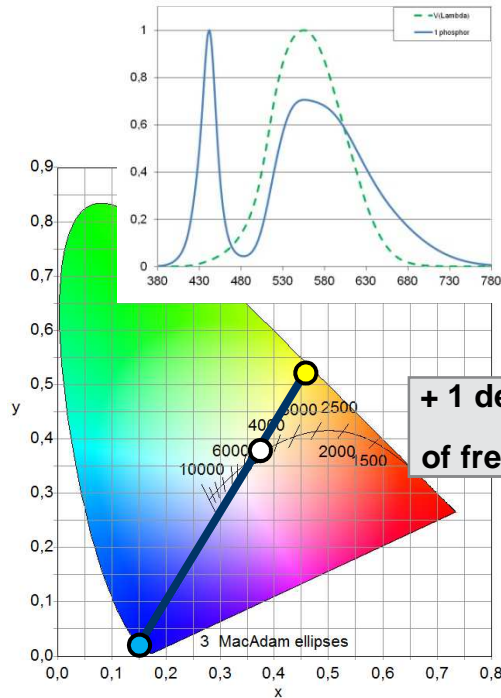
Major Phosphor parameters

Spectral emission	$\lambda(\text{dom})$	dominant emission wavelength
	FWHM	spectral emission width
	Cx, Cy	color coordinate
Spectral absorption	λ_{Rmin}	wavelength position of remission minimum
	$R_{\lambda_{\text{ex}}}$	remission at excitation wavelength
Phosphor efficiency	QE	quantum efficiency
Grain characteristic	Size	nm-particle → LED eff. ↓, homogeneous scattering, uniform color
		μm-particle → LED eff. ↑, forward scattering, color dependant on angle
Chemical composition (example YAG)	Shape	spherical, rough etc. impact on eff. and conversion
	c(Ce)	low concentration → less t-quenching, less absorption, green shift
	c(Ga)	high conc. → green shifted emission and absorption
	c(Gd)	high conc. → red shifted emission

→ All parameters are important at the same time

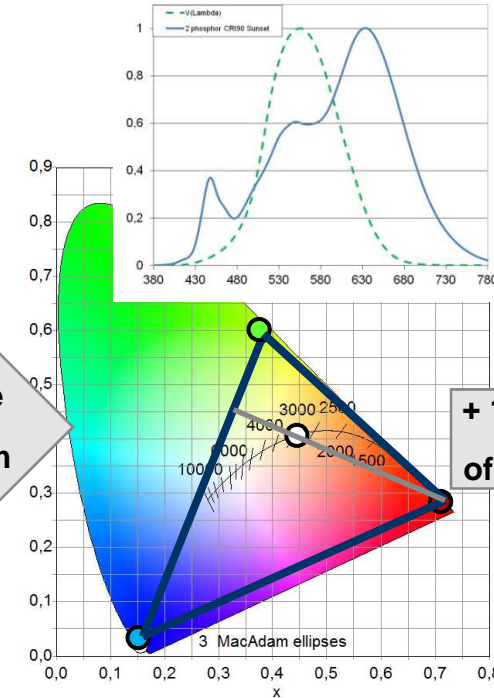
Multi-Phosphor Solutions: 1, 2, 3 or even more?

1 Phosphor: Cold White



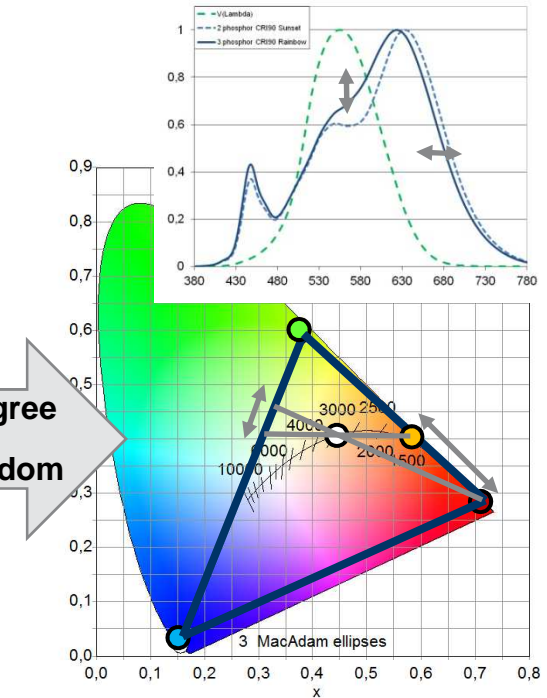
- 1 CCT on Planck

2 Phosphors: Warmwhite



- adjustable CCT
- CRI fixed with CCT

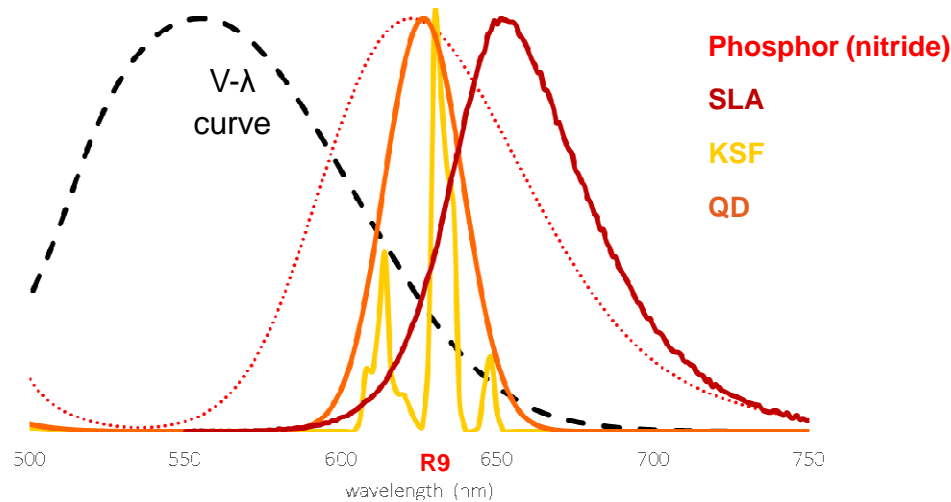
3 Phosphors: LED3



- adjustable CCT
 - fine-tuned CRI
- gain of efficacy

⇒ Better fine-tuning with more phosphors

Outlook: narrow-band converters



SLA = $\text{SrLiAl}_3\text{N}_4:\text{Eu}$
KSF = potassium fluorosilicate $\text{K}_2\text{SiF}_6:\text{Mn}^{4+}$
QD = Quantum Dots
(nanoscale semiconductor particles)

- SLA: wavelength too long
- KSF: low absorption, long decay time
- QD: flexible wavelength and FWHM but limited stability

Summary

- Many different material classes are used and needed for different LED packages
- Each material has to fulfill many requirements at the same time
- **Interaction of the materials and the LED chip is key for LED development**

